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Subject: Engineering Chemistry Subject Code: 17CHE12/22 Duration: 3 Hours Max. marks: 100 Note: Answer any FIVE full questions, choosing one full question from each module Module -I 1 a. What is single electrode potential? Derive the Nernst equation for s ingle electrode potential . (7 marks) b. What are concentration cells? The cell potential of Cu concentration cell Cu/CuSO₄ (0.005M ...

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Subject Code 17CHE12/17CHE22 IA Marks 40 Number of Lecture Hours/Week 04 Exam Marks 60 Total Number of Lecture Hours 50 Exam Hours 03 CREDITS - 04 Course objectives: To provide students with knowledge of engineering chemistry for building technical competence in industries, research and development in the following fields • Electrochemistry & Battery Technology. • Corrosion & Metal ...

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Visvesvaraya Technological University Jnana Sangama, Belagavi — 590 018. Prof. Satish Annigeri Phone . (0831)2498133 B.E. 2018 B.E. 2018 Technical English —4 | 18CHE12/22 Engineering Chemistry Monday, 07/09/2020 2.00 pm 5.00 pm Monday, 14/09/2020 2.00 pm 5.00 pm Friday, 11/09/2020 2.00 pm — 5.00 pm 19/09/2020 Saturday, 5.00 pm 2.00 pm 44 Registrar (Evaluation) To, The Principals of all ...

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